



## 1 Decay Scheme

Mn-54 decays by electron capture (99.9997(3) %) directly to the 834.855-keV excited level (2+) of Cr-54 and by very weak electron capture and beta plus transitions to the ground state (0+) of Cr-54.

*Le manganèse 54 décroît par capture électronique (99,9997 (3) %) vers le niveau excité à 834,855 keV du chrome 54 et par capture électronique et transition bêta plus vers le niveau fondamental du chrome 54.*

## 2 Nuclear Data

$$T_{1/2}({}^{54}\text{Mn}) : 312,19 \quad (3) \quad \text{d}$$

$$Q^+({}^{54}\text{Mn}) : 1377,2 \quad (10) \quad \text{keV}$$

### 2.1 Electron Capture Transitions

	Energy (keV)	Probability (%)	Nature	lg <i>ft</i>	<i>P<sub>K</sub></i>	<i>P<sub>L</sub></i>	<i>P<sub>M</sub></i>
ε <sub>0,1</sub>	542,3 (10)	99,9997 (3)	Allowed	6,17	0,8896 (17)	0,0948 (14)	0,0150 (6)
ε <sub>0,0</sub>	1377,2 (10)	0,0003 (3)	Unique 2nd Forbidden	>13,9	0,8908 (16)	0,0938 (13)	0,0148 (6)

### 2.2 β<sup>+</sup> Transitions

	Energy (keV)	Probability (%)	Nature	lg <i>ft</i>
β <sub>0,0</sub> <sup>+</sup>	355,2 (10)	<0,00000057	Unique 2nd Forbidden	>13,9

### 2.3 Gamma Transitions and Internal Conversion Coefficients

	Energy (keV)	P <sub>γ+ce</sub> (%)	Multipolarity	α <sub>K</sub> (10 <sup>-4</sup> )	α <sub>L</sub> (10 <sup>-4</sup> )	α <sub>M</sub> (10 <sup>-4</sup> )	α <sub>T</sub> (10 <sup>-4</sup> )
γ <sub>1,0</sub> (Cr)	834,855 (3)	99,9997 (3)	E2	2,22 (4)	0,206 (3)	0,0271 (4)	2,45 (4)

## 3 Atomic Data

### 3.1 Cr

ω <sub>K</sub>	:	0,289	(5)
ω̄ <sub>L</sub>	:	0,0045	(9)
n <sub>KL</sub>	:	1,508	(5)

#### 3.1.1 X Radiations

	Energy (keV)	Relative probability
X <sub>K</sub>		
Kα <sub>2</sub>	5,40557	50,91
Kα <sub>1</sub>	5,41479	100
Kβ <sub>1</sub>	5,94677	} 20,31
Kβ <sub>5</sub> ''	5,987	
X <sub>L</sub>		
Lℓ	0,5003	
Lα	0,5729 - 0,57695	
Lη	0,5102	
Lβ	0,57515 - 0,69748	
Lγ	0,58496 - 0,58496	

#### 3.1.2 Auger Electrons

	Energy (keV)	Relative probability
Auger K		
KLL	4,554 - 4,794	100
KLX	5,206 - 5,412	26,92
KXY	5,841 - 5,985	1,913
Auger L	0,4 - 0,7	2,91

## 4 Electron and Positron Emissions

		Energy (keV)	Electrons (per 100 disint.)
e <sub>AL</sub>	(Cr)	0,4 - 0,7	143,0 (6)
e <sub>AK</sub>	(Cr)		
	KLL	4,554 - 4,794	} 63,3 (5)
	KLX	5,206 - 5,412	
	KXY	5,841 - 5,985	
ec <sub>1,0 K</sub>	(Cr)	828,866 (3)	0,0222 (4)
ec <sub>1,0 L</sub>	(Cr)	834,16 - 834,281	0,00206 (3)
$\beta_{0,0}^+$	max:	355,2 (10)	} <0,00000057
	avg:	182	

## 5 Photon Emissions

### 5.1 X-Ray Emissions

		Energy (keV)	Photons (per 100 disint.)	
XL	(Cr)	0,5003 - 0,69748	0,65 (13)	
XK $\alpha_2$	(Cr)	5,40557	7,65 (14)	} K $\alpha$
XK $\alpha_1$	(Cr)	5,41479	15,02 (27)	
XK $\beta_1$	(Cr)	5,94677	} 3,05 (7)	K' $\beta_1$
XK $\beta_5''$	(Cr)	5,987		

### 5.2 Gamma Emissions

		Energy (keV)	Photons (per 100 disint.)
$\gamma^\pm$		511	<0,00000114
$\gamma_{1,0}$ (Cr)		834,848 (3)	99,9752 (5)

## 6 Main Production Modes

$\left\{ \begin{array}{l} \text{Fe} - 54(\text{n,p})\text{Mn} - 54 \\ \text{Possible impurities: Cr} - 51, \text{Fe} - 55, \text{Fe} - 57 \end{array} \right.$

- { Cr – 53(d,n)Mn – 54
- { Possible impurities: V – 48, Mn – 52
- V – 51( $\alpha$ ,n)Mn – 54
- Cr – 52( $\alpha$ ,d)Mn – 54
- Cr – 54(p,n)Mn – 54
- Cr – 52(t,n)Mn – 54
- Fe – 54(t,He – 3)Mn – 54

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